



Cell: 083 263 1026 WhatsApp: 072 039 6116 PRN:0900040063843 Email: pedro@eyesalive.co.za Website: www.eyesalive.co.za Practice or consultation hours: 09:30 - 16:30 Mon to Fri by appointment only Postal Address: PO BOX 1266, Olivedale, 2158. Randburg: 32 Ebbehout St, Cnr Mahogany St, Sharonlea. Centurion: Suite 115, Centurion Eye Hospital, Unitas Lifestyle Management Park 4, Clifton Ave, Centurion.



any patients wonder why Eyes Alive work the way we do. Surely an Ocular Prosthesis can be made in less time. In the old days, eyes were made in a day.

Over the last few years, one of the biggest concerns in the manufacturing of an Ocular Prosthetic is **Free Monomer in ocular prosthetics after curing the prosthesis.** Globally, lots of research and debate exists around this matter. In essence, there is agreement that the old philosophy of short single-stage curing is no longer appropriate.

When a prosthesis is cured too quickly, it potentially leaves free monomer trapped inside the prosthesis. With Free monomer in the prosthesis, the patient will often experience symptoms like dry-eye, excessive discharge, and overall discomfort.

Some patients interpret this discharge as an infection in the socket and visit their local GP or Ophthalmologist for antibiotic treatment. The symptoms often clear up and repeat soon after the prescribed medication course ends. This confirmed the problem is more likely related to the prosthesis than the socket, and the patient should visit their Ocularist.

At Eyes Alive, we made the decision that our patient's wellbeing is paramount, and therefore we manufacture our eyes without exception using a slower multistage process, which is in line with globally accepted best practices. Although this takes much longer to manufacture and fit an eye, and costs Eyes alive a lot more resources, the patients overall report a happier experience wearing their prosthesis.

A quick look at why practitioners should be worried about free monomer or residual monomer in the prosthesis.

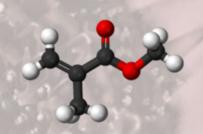


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### Monomer is a complex molecule

Wikipedia defines the molecule as follows:

| Properties  |  |
|---|--|
| Chemical formula  | C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>   |
| Molar mass  | 100.117 g·mol <sup>−1</sup>  |
| Appearance  | Colourless liquid  |
| <u>Odor</u>   | acrid, fruity <sup>[1]</sup>   |
| <u>Density</u>  | 0.94 g/cm <sup>3</sup>   |
| Melting point   | –48 °C (–54 °F; 225 K)   |
| Boiling point   | 101 °C (214 °F; 374 K)   |
| Solubility in water                                     | 1.5 g/100 ml   |
| log P   | 1.35 [2]   |
| Vapor pressure  | 29 mmHg (20°C) <sup>[1]</sup>  |
| Magnetic susceptibility (χ)                             | -57.3·10 <sup>-6</sup> cm <sup>3</sup> /mol  |
| <u>Viscosity</u>  | 0.6 c <u>P</u> at 20 °C  |
| Structure   |  |
| Dipole moment   | 1.6–1.97 <u>D</u>  |
| Hazards   |  |
| Main <u>hazards</u>                                     | Flammable  |
| Safety datasheet  | <i>See: <u>data page</u><br/>Methyl methacrylate MSDS</i>  |
| <u>Flashpoint</u>                                       | 2 °C (36 °F; 275 K)  |
| Autoignition<br>temperature                             | 435 °C (815 °F; 708 K)   |
| Explosive limits  | 1.7%-8.2%[1]   |
| Lethal dose or concentration ( <i>LD</i> , <i>LC</i> ): |  |
| LC <sub>50</sub> (median concentration)                 | 18750 ppm (rat, 4 hr)<br>4447 ppm (mouse, 2 hr)<br>3750 ppm (rat)<br>4808 ppm (mammal) <sup>[3]</sup>                      |
| LC <sub>Lo</sub> (lowest published)                     | 4400 ppm (rat, 8 hr)<br>4400 ppm (rabbit, 8 hr)<br>4207 ppm (rabbit, 4.5 hr)<br>4567 ppm (guinea pig, 5 hr) <sup>[3]</sup> |
| US health exposure limits ( <u>NIOSH</u> ):             |  |
| PEL (Permissible)                                       | TWA 100 ppm (410 mg/m <sup>3</sup> ) <sup>[1]</sup>  |
| REL (Recommended)                                       | TWA 100 ppm (410 mg/m <sup>3</sup> ) <sup>[1]</sup>  |
| IDLH (Immediate danger)                                 | 1000 ppm <sup>[1]</sup>  |



A quick visit to Wikipedia and the toxic value of monomer is evident.

In rat studies, where rats are exposed to the vapours of Monomer, death occurs in less than two hours.

Monomer's flashpoint or, the temperature that it evaporates varies between sources, but it is generally accepted to be between 2°C and 12°C.

This, in turn, means any monomer trapped in the prosthesis will leach out into the socket for the life of the prosthesis, due to the body being at a constant 37°C.

(Note that most of the literature indicates that the critical period is the first 48 hours from manufacturing, after which the residual levels seem to drop and are quite low.

However low, these levels may be, there remain associated health hazards).

The eye socket is similar to the mouth, where the same dental materials are used, in the sense that it has wet tissue.

Wet tissue assists the monomer to leach, and the free monomer in the body fluids is then easily absorbed into the blood stream and circulated through the body.

Studies indicate that the first 24-48 hours is most significant.

Therefore, if we extend our curing process to include an additional 24-hours, we can almost guarantee the best result for our patients, with the lowest possible monomer dosing.





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### Multistage curing

What is multistage curing?

The old philosophy was to place the acrylic into water and heat it immediately in one step to 100 degrees. This for reasons explained below is not ideal, so the curing unit needs to be able to raise the temperature in a controlled fashion, in steps or ramps, i.e. take the water with the prosthesis up to a lower temp, say 53 degrees, over a predetermined timeframe, of say 15 minutes, then hold at that temp for some time before it raised again to a higher temperature. This allows the heat to penetrate the full thickness of the mould, which houses the prosthesis during the curing.

Although the acrylic manufacturer specifications, in some instances, only call for a 15-minute cure at 100 degrees. This specification is designed to achieve the maximum material strength of the acrylic; however, there is enough documented scientific evidence to show this is not good enough for patients as it may not remove the free monomer.

Another concern of some researchers is that when you place a sample of material into a curing vessel and take it straight to 100 degrees, you get what is referred to as case hardening or shell hardening.

What this means is that if your curing vessel is switched on and allowed to heat up at the maximum rate of the element, the outside of the prosthesis heats up much faster than the inside section, thus as the outside of the prosthesis cures, and it traps some of the monomer in the prosthesis before the inner part reaches the temp required to cure.

At Eyes Alive, our curing programme, the first stage will be taking the water from ambient temp or room temp to  $53^{\circ}$ C, and we programme this to take 15 minutes.

By forcing the element to work slower, we guarantee that the entire prosthesis cures at a uniform temperature throughout. This, in turn, minimises the possibility of Shell Hardening and trapping monomer within the prosthesis.

The controller then holds the temp at  $53^{\circ}$ C for a time before starting the next ramp which eventually takes our temp up to  $97^{\circ}$ C.

This process is referred to in the thermal industry as Ramp Control.

Below, I will describe the full heat profile we use to cure eyes at Eyes Alive.

Our heat profile has been developed and tweaked over a number of years, taking cognisance of global best practices.

This curing process must happen for each stage of the eye, so you can understand that most curing is done overnight, and the eye gets cured between each appointment with the patient, which means the cumulative time to make an eye is a minimum of three days.



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The GRAND PRIZE, is we don't have patients with sensitivity to monomer. We have, however, had several patients where their prosthesis was made elsewhere, and they have approached us with a complaint of irritations to the socket. In most cases, if we can run their prosthesis through our multistage curing process, and these symptoms are a thing of the past.

### Establishing a good ramp profile/heat profile for curing an eye.

In our practice, the heat profile or ramp profile we use has been developed to what we believe gives the patient maximum security that they will not likely experience any socket irritations from the residual monomer.

Our programme looks like this:

Ramp 1 - Ambient to 53°C over 15 minutes

- Ramp 2 Hold at 53°C for 15 minutes
- Ramp 3 53°C to 73°C over 15 minutes
- Ramp 4 Hold at 73°C for 4 hours

Ramp 5 - 73°C to 97°C over 15 minutes

- Ramp 6 hold at 97°C for 5 hours
- Ramp 7 97°C back to ambient temp in its own time



This reiterates why manufacturing an eye in a day does not happen in our practice. In fact, curing each stage in this manner forces manufacturing to take, at the very least, three days per patient, excluding the variance, of needing to make adjustments and where necessary allow the patient to reduce swelling, wear the model to confirm comfort over a few days or other unforeseen inconveniences.



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From the above explanation, we can be guaranteed that we have reached maximum tensile strength, and the acrylic will be stable and well within the manufacturer's specifications.

However, we can't be sure of what free monomer is left in the prosthesis in the form of the residual monomer or free monomer that can leach out over the rest of the prosthesis' lifespan. So, looking at the literature on best practices to remove free monomer. It is suggested to let the final prosthetic eye soak for a further 24 hours in distilled water at 53°C.

Ideally one must use distilled water, as there will be zero dissolved solids and other contaminants in the water, which will allow maximum leaching. Thereafter, the prosthesis is polished to a high gloss and fitted to the patient. All this work enforces Eyes Alive's commitment to service excellence and customer satisfaction.

Resources:

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Canan BURAL12, Gunnur DENIZ3 4, Gulsen BAYRAKTAR5 Faculty of Dentistry - Department of Removable Prosthodontics - Capa, 34390 - Istanbul -Turkey

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1University of Niš, Medical Faculty, Bulevar dr Zorana Đinđića 81, Niš, Serbia 2University of Niš, Faculty of Technology, Bulevar Oslobođenja 124, Leskovac, Serbia 3University of Niš, Faculty of Mechanical Engineering, Aleksandra Medvedeva 14, Niš, Serbia

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Product identifier Product name Vertex Monomer Type A Product description Monomer based on Methyl Methacrylate Alternative names Vertex Rapid Simplified, Vertex Regular, Vertex Implacryl, Vertex Implacryl Cold, Vertex Castapress, Vertex Castavaria, Vertex TCA, Vertex Teeth Material, Vertex BasiQ 20, Platinum HI, J-Cryl+. SDS ID.:MMOA201504UK Page 1 of 9 (Date: 12/15)

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He was editor of TandartsPraktijk between 1990 and 2010.

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